pH-Dependent Conformation, Dynamics, and Aromatic Interaction of the Gating Tryptophan Residue of the Influenza M2 Proton Channel from Solid-State NMR

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ABSTRACT The M2 protein of the influenza virus conducts protons into the virion under external acidic pH. The proton selectivity of the tetrameric channel is controlled by a single histidine (His37), whereas channel gating is accomplished by a single tryptophan (Trp41) in the transmembrane domain of the protein. Aromatic interaction between these two functional residues has been previously observed in Raman spectra, but atomic-resolution evidence for this interaction remains scarce. Here we use high-resolution solid-state NMR spectroscopy to determine the side-chain conformation and dynamics of Trp41 in the M2 transmembrane peptide by measuring the Trp chemical shifts, His37-Trp41 distances, and indole dynamics at high and low pH. The interatomic distances constrain the Trp41 side-chain conformation to trans for \( \chi_1 \) and 120–135° for \( \chi_2 \). This \( \phi \theta \) rotamer points the Ne1-Cr2-Cz2 side of the indole toward the aqueous pore. The precise \( \chi_1 \) and \( \chi_2 \) angles differ by ~20° between high and low pH. These differences, together with the known changes in the helix tilt angle between high and low pH, push the imidazole and indole rings closer together at low pH. Moreover, the measured order parameters indicate that the indole rings undergo simultaneous \( \chi_1 \) and \( \chi_2 \) torsional fluctuations at acidic pH, but only restricted \( \chi_1 \) fluctuations at high pH. As a result, the Trp41 side chain periodically experiences strong cation-π interactions with His37 at low pH as the indole sweeps through its trajectory, whereas at high pH the indole ring is further away from the imidazole. These results provide the structural basis for understanding how the His37-water proton exchange rate measured by NMR is reduced to the small proton flux measured in biochemical experiments. The indole dynamics, together with the known motion of the imidazolium, indicate that this compact ion channel uses economical side-chain dynamics to regulate proton conduction and gating.

INTRODUCTION

The influenza A M2 protein forms a homo-tetrameric proton channel that acidifies the virus after endocytosis (1–3) and maintains the high pH of the Golgi network for proper hemagglutinin function (4). M2 also mediates virus budding from the host cell by causing membrane scission (5). The proton channel is activated by low pH of the external environment, which is presented by the endosome during endocytosis. A conserved HxxxW motif is responsible for unidirectional proton conduction from the exterior into the virion. In this HxxxW motif, His37 is now known from solid-state NMR data to exchange protons with water molecules at a rate of \( 10^7 \) s\(^{-1} \), facilitated by small-angle ring reorientations on the same timescale (6,7). The histidine side chain adopts the trans-trans conformation for the \( \chi_1 \) and \( \chi_2 \) torsion angles, such that Nδ1 and Ne2 point toward the N- and C-termini of the channel, respectively, primed to accept and release protons. The single-channel proton conductivity is the highest when three out of four histidines of the tetramer are protonated, which occur with a pK\(_a\) of 4.9 in cholesterol-rich virus-mimetic lipid membranes (7). The water-His37 proton transfer is facilitated by an extensive network of imidazole-water hydrogen bonds at acidic pH (6,8), but the hydrogen-bonding network is incomplete at high pH, when the channel is closed.

Whereas His37 is the proton-selective residue, Trp41 is the gating residue, blocking diffusion of protons from inside the virus but not from the outside (9). Electrophysiology data show that when pH\(_{\text{out}}\) is lower than pH\(_{\text{in}}\), there is robust inward proton flux, which is much larger than the outward current when the situation is reversed. This asymmetry is lost when Trp41 is replaced by other amino-acid residues except Tyr, which also has an electron-rich aromatic ring capable of forming cation-π interactions with His37. These results suggest that when the Trp41 gate is closed, protons cannot rapidly access His37 from inside the virus, thus outward flux is slow under low pH\(_{\text{in}}\) and high pH\(_{\text{out}}\). Despite these functional data on the role of Trp41 in the channel activity, high-resolution structural information about pH-dependent His37-Trp41 aromatic interaction is still scarce, and the Trp41 side-chain conformation is still unresolved. Various high-resolution structures of M2 domains showed divergent Trp41 rotamers that placed the indole ring at varying positions relative to the pore (Table 1). Most structural models found the \( \chi_1 \) angle to be trans \((t)\) but differed on whether the \( \chi_2 \) angle is positive or negative 100°. 19F spin diffusion NMR experiments of 5-19F (H3)3-labeled Trp41 in the M2 transmembrane peptide (M2TM) (10) found nearest-neighbor 19F-19F distances of ~11 Å at both high and low pH, suggesting a \( \phi \theta \) rotamer, which points the five-membered nitrogen-containing pyrrole ring...
toward the channel pore (Fig. 1 a). Crystal structures of M2TM at pH 6.5 and pH 5.3 also concluded the φ90 rotamer (11,12). In comparison, an earlier solid-state NMR measurement of the His37 Ne2-Trp41 Cγ distance in M2TM found a distance upper limit of 3.9 Å, which was interpreted to constrain the Trp rotamer to t-105 (x1 = 180°; x2 = −105°) (13) (Fig. 1 b). The negative x2 angle points the six-membered benzene ring toward the center of the channel, occluding the pore. Solution NMR structures of a combined transmembrane (TM) and cytoplasmic helix domain of wild-type and mutant M2 proteins also concluded the Trp41 rotamer to be t-105 based on the side-chain J-couplings, residual dipolar couplings, and nuclear Overhauser effects (14,15). Finally, a 19F NMR study of the lineshapes of 6-19F (Hπ2) Trp41 in M2TM (16) found a nearest-neighbor 19F-19F distance of ~8.0 Å at low pH and ~3.2 Å at high pH, which led to the proposal of a positive χ2 of +100°, but a negative χ1 of −100° and −50° for the high- and low-pH states, respectively (Fig. 1 c).

Direct experimental evidence of His37-Trp41 aromatic interactions so far mainly comes from ultraviolet resonance Raman spectra, which showed Trp41 intensity changes at certain wavenumbers at low pH (17). These changes were attributed to cation-π interactions between the imidazolium and the indole rings, whereas environmental hydrophobicity and indole hydrogen bonding were ruled out. A recent molecular dynamics simulation that modeled the HxωxW segment led to the proposal that cation-π interactions at low pH disrupted a low-barrier hydrogen bond between a cationic and neutral histidine (18), thus activating the channel. However, measured chemical shifts of the protons bonded to the imidazole nitrogens indicate regular His-water hydrogen bonds (8) rather than direct His-His hydrogen bonds.

In this work, we investigate the conformation and dynamics of Trp41 in M2TM by measuring the Trp41 chemical shifts, His37-Trp41 distances, and indole dynamics as a function of pH. We show that Trp41 adopts the φ90 configuration at both high and low pH, but the exact χ1 and χ2 angles differ by ~20°. These torsion angle differences, together with the known increase in the helix tilt angle at low pH, push the imidazole of one helix toward the indole of the neighboring helix at low pH. Moreover, measured order parameters indicate that the indole ring undergoes Gaussian fluctuations around both the χ1 and χ2 bonds at low pH, which further promote His-Trp interaction. These results give fresh insight into the coordinated motion between His37 and Trp41 that regulates proton conduction and channel gating.

MATERIALS AND METHODS

Membrane peptide samples

The TM segment (residues 22–46: SSDPLVVAASII GLHILILWILDRL) of the Udorn strain of the influenza A M2 protein was synthesized by Primm-Biotech (Cambridge, MA). Two isotopically labeled M2TM peptides were synthesized, one containing uniformly 13C-, 15N-labeled Leu40 and Trp41 side chains are depicted. Note the shortest distances between these two residues are between two adjacent helices rather than from the same helix. (Right column) Top view of the four helices and the Trp41 sidechains. (a) φ90 rotamer (x1 = 180°; x2 = +90°). (b) t-105 rotamer (x1 = 180°; x2 = −105°). (c) m95 rotamer (x1 = −60°; x2 = +95°).

FIGURE 1 Several Trp41 rotamers that have been proposed in the literature using various experimental techniques (Table 1). Population percentages for φ-helices from the Penultimate Rotamer Library are shown (42).

TABLE 1 Trp41 rotamers in various structures of the influenza M2 peptides, solved at different pH, in different membrane-mimetic solvents, using different protein constructs and biophysical methods

<table>
<thead>
<tr>
<th>PDB ID</th>
<th>pH</th>
<th>Lipid/detergent</th>
<th>Method</th>
<th>Construct</th>
<th>Trp41 (x1, x2)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1NYJ</td>
<td>7.0</td>
<td>DMPC</td>
<td>SSNMR</td>
<td>22–46</td>
<td>180°, −105°</td>
</tr>
<tr>
<td>2H95</td>
<td>8.8</td>
<td>DMPC</td>
<td>SSNMR</td>
<td>22–46</td>
<td>−100°, 110°</td>
</tr>
<tr>
<td>2L0J</td>
<td>7.5</td>
<td>DOPE/DOPE</td>
<td>MD, SSNMR</td>
<td>22–62</td>
<td>180°, −70°</td>
</tr>
<tr>
<td>2KAD</td>
<td>7.5</td>
<td>DLPC</td>
<td>SSNMR</td>
<td>22–46</td>
<td>180°, 90°</td>
</tr>
<tr>
<td>2KQT</td>
<td>7.5</td>
<td>DMPC</td>
<td>SSNMR</td>
<td>22–46</td>
<td>180°, 90°</td>
</tr>
<tr>
<td>3C9J</td>
<td>5.3</td>
<td>Octylglucoside</td>
<td>X-ray</td>
<td>22–46, G34A</td>
<td>180°, 90°</td>
</tr>
<tr>
<td>3LBW</td>
<td>6.5</td>
<td>Octylglucoside</td>
<td>X-ray</td>
<td>22–46, G34A</td>
<td>180°, 90°</td>
</tr>
<tr>
<td>2RLF</td>
<td>7.5</td>
<td>DHPC</td>
<td>Solution NMR</td>
<td>18–60</td>
<td>160°, −120°</td>
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<td>7.5</td>
<td>DHPC</td>
<td>Solution NMR</td>
<td>18–60, V27A</td>
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<td>7.5</td>
<td>DHPC</td>
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<td>18–60, S31N</td>
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<td>7.5</td>
<td>DHPC</td>
<td>Solution NMR</td>
<td>18–60, S31N</td>
<td>180°, −115°</td>
</tr>
</tbody>
</table>

DHPC, Dihexanoylphosphatidylcholine.
molar ratio of 21:21:28:30%, whereas the VM+ membrane consists of POPC (1-palmitoyl-2-oleoyl-sn-glycero-3-phosphocholine), POPE (1-palmitoyl-2-oleoyl-sn-glycero-3-phosphoethanolamine), SM, and Chol at a molar ratio of 25.6:25.6:25.6:25.6:23%. The VM+ membrane is more dynamic than the VM membrane due to the presence of unsaturated phospholipids.

For both membranes the lipids and cholesterol were dissolved in chloroform, SM was dissolved in chloroform/methanol solution, and all components were combined at the desired molar ratios. The lipid mixture was dried under a stream of nitrogen gas, redissolved in cyclohexane, and lyophilized overnight. The resulting dry lipid powder was resuspended in the appropriate buffer, vortexed, and subjected to 8–10 cycles of freeze-thawing to produce uniform vesicles. Lipid solutions were prepared at two different pH: the pH 8.5 sample used a 10 mM Tris buffer (10 mM Tris, 1 mM EDTA, 0.1 mM NaNO3) whereas the pH 4.5 sample used a 10 mM citric acid buffer (10 mM citric acid/sodium citrate, 1 mM EDTA, and 0.1 mM NaNO3).

M2TM was reconstituted into the lipid membrane by detergent dialysis. The peptide was dissolved in an octyl-β-D-glucoside (OG) solution that is below the critical micelle concentration of 7.3 mg/mL at room temperature (21). The mixed solution was shaken for 2 h and dialyzed for three days at room temperature with six buffer changes to remove the detergent. The LW-M2TM sample was reconstituted in the VM+ membrane, whereas the GHW-M2TM sample was reconstituted in the VM membrane. The peptide/lipid molar ratio was 1:15 for all samples. The dialyzed proteoliposomes solutions were centrifuged at 55,000 rpm at 5°C for 4 h to obtain homogeneous membrane pellets, which were spun into 4 mm MAS rotors for solid-state NMR experiments.

### Solid-state NMR experiments

All solid-state NMR (SSNMR) experiments were performed on a Bruker DSX-400 MHz (9.4 T) spectrometer (Bruker Biospin Billerica, MA) using two 4-mm MAS probes tuned to $^{1}H$/$^{13}C$/$^{15}N$ and $^{1}H$/$^{19}F$ frequencies. Typical radio frequency field strengths were 50 kHz for $^{13}C$, 42 kHz for $^{15}N$, 50 kHz for $^{19}F$, and 62–83 kHz for $^{1}H$. $^{13}C$ chemical shifts were referenced to the α-Gly carbonyl at 176.465 ppm on the tetramethylsilane scale, whereas $^{15}N$ chemical shifts were referenced to the $^{15}N$ signal of N-acetylvaline at 122.0 ppm on the lipid ammonia scale. $^{19}F$ chemical shifts were referenced to the $^{19}F$ signal of Teflon at −122.0 ppm.

One-dimensional (1D) $^{13}C$ cross-polarization spectra were measured under 7 kHz MAS between 243 K and 303 K. One-dimensional $^{13}C$ double-quantum-filtered (DQF) spectra were measured at 303 K under 7 kHz MAS. Two-dimensional (2D) DQF $^{13}C$-$^{13}C$ correlation spectra were measured at 273 K under 7 kHz MAS. To determine motional order parameters, we measured $^{1}C$-$^{1}H$ dipolar couplings using the dipolar chemical-shift correlation experiment (DIPSHIFT) under 4.4 kHz MAS at 303 K. $^{1}H$ homonuclear decoupling was achieved using the MREV-8 sequence with a 90° pulse length of 3.5 μs. An MREV-8 scaling factor of 0.47 and a rigid-limit coupling of 22.7 kHz were used to calculate the Trp41 order parameters. Ce1-Nε1 and Cβ1-Nε1 dipolar couplings were measured using the rotational-echo double-resonance (REOR) experiment (22) under 7 kHz MAS at 303 K. To calculate the C-N dipolar order parameter, a rigid-limit coupling of 1.15 kHz was used, which was verified on amino-acid tryptophan.

$^{13}C$-$^{19}F$ REDOR experiment was used to measure $^{13}C$-$^{19}F$ distances between $^{13}C$-labeled His37 and $^{19}F$-Trp31 in the GHW-M2TM samples at pH 8.5 and 4.5. The REDOR pulse sequence utilized composite 90° 225° 315° $^{19}F$ pulses to achieve broad bandwidth inversion of the $^{19}F$ polarization (23). A soft Gaussian $^{1}C$ 180° pulse was applied in the middle of the REDOR period to refocus the $^{1}C$ chemical shift and the $^{13}C$-$^{13}C$ scalar coupling in the uniformly $^{13}C$-labeled His37 (24, 25). For the low-pH sample, MAS frequencies of 3.0, 3.3, and 6.6 kHz were used to detect the aromatic $^{13}C$ signals of His37 without overlap from the CO sidebands. For the high-pH sample, the REDOR experiment was conducted under 3.3 and 5.0 kHz MAS. All $^{13}C$-$^{19}F$ REDOR spectra were measured at 243 K and the mixing times varied from 1.2 to 9.6 ms, with 20,000–70,000 scans of signal averaging per spectrum.

The centerband-only-detection of exchange (CODEX) experiment (26, 27) was used to measure interhelical Trp$^{31}$-Trp$^{41}$ distances. The experiments were conducted under 8 kHz MAS at ~230 K. Two rotor periods of $^{19}F$ π-pulses spaced at every half a rotor period was applied before and after a longitudinal mixing time to recouple the $^{19}F$ chemical shift anisotropy. After the second π-pulse train, a second longitudinal mixing time was applied to factor out $^{19}F T_1$ relaxation. Analogous to the REDOR experiments, the CODEX experiments were conducted in a pairwise fashion, with a control ($S_0$) and dephasing ($S$) experiment, and normalized intensity ($S_0/S$) is fit to obtain distances. CODEX mixing times of 1 ms to 2 s were used.

### Data analysis and simulation

The error bars for the $^{13}C$-$^{19}F$ REDOR data points were propagated from the experimental signal/noise ratios. The $^{13}C$-$^{19}F$ REDOR $S_0/S$ values as a function of mixing time were first simulated using the program SIMPSON (28), assuming two-spin geometry. Best-fit distances were obtained by minimizing the root-mean-square deviations (RMSDs) between the simulated and experimental intensities. After the Trp$^{41}$ $(χ_1, χ_2)$ rotamers were determined, we used the SPINEVOLUTION program (29) to conduct model-dependent five-spin (one $^{13}C$ and four $^{19}F$ spins) simulations for each possible rotamer. PDB files were generated for each Trp$^{41}$ rotamer, then the coordinates for each Trp$^{41}$ $^{13}C$ and $^{19}F$ sites were provided to the SPINEVOLUTION program for use in the five-spin simulations.

The $^{19}F$ CODEX exchange data were fit using a MATLAB program (The MathWorks, Natick, MA) that employs the exchange-matrix formalism to treat spin diffusion in the four-spin system. The 4 × 4 exchange matrix contains rate constants that are proportional to an overlap integral and the square of the intermolecular dipolar couplings. Based on previous model compound results, the overlap integral value is 37 μs (10, 30) under our experimental conditions. Best-fit distances were obtained by minimizing the RMSD between the calculated and experimental CODEX intensities.

DIPSHIFT curves are simulated using a FORTRAN program. A phenomenological $T_2$ relaxation time was applied to the best-fit simulated curves to reproduce the observed asymmetry in the time signal.

### Analysis of Trp$^{41}$ rotameric conformation

To model the side-chain conformation of Trp$^{41}$ from the measured His-Trp and Trp-Trp distances, we first considered the choice of the backbone structure. For the high-pH data, we used the distance-constrained solid-state NMR structure (PDB:2KQT) and the 1.65-Å crystal structure (PDB:3LBW) as backbone structural models, because they have similar helix tilt angles. For the low-pH data, we used the pH 5.3 crystal structure (PDB:3C9J) and the DLPC (dilauroylphosphatidylcholine)-based solid-state NMR structural model (PDB:2KAD) (31, 32). These two backbone structures resemble each other in having a helix tilt angle of 35–38°, which is consistent with the helix orientation measured using oriented-membrane solid-state NMR experiments (13, 33). The DLPC-based solid-state NMR structure has C4 symmetry, which simplifies the rotamer analysis, whereas the low-pH crystal structure (PDB:3C9J) has significant asymmetry.

In all distance analyses, the His$^{37}$ side chain was fixed to the $n$ rotamer $(χ_1 = χ_2 = 180°)$ determined by SSNMR and crystallography (6, 11). The Trp$^{31}$ $χ_1$ and $χ_2$ angles were varied and the His-Trp $^{13}C$-$^{13}C$ distances and Trp-Trp intermolecular $^{13}C$-$^{13}C$ distances were measured and compared to the experimental data. In a first round of analysis, the Trp$^{31}$ $χ_1$ and $χ_2$ angles were manually varied in 30° steps in the program CHIMERA (34) to obtain a coarse-grained map of the possible Trp$^{31}$ rotamers that agree with the experimental data. These results were then refined in a second round of analysis, where the software YASARA (35) was used to tabulate the...
Calculation of motionally averaged dipolar couplings in the Trp$^{41}$ side chain

The motional amplitudes around the Ca-Cβ and Cβ-Cy bonds of Trp$^{41}$ were reflected in the motionally averaged C-H and C-N dipolar couplings, i.e., order parameters. To constrain the motional geometry, we calculated the order parameters of the indole ring using a Gaussian biaxial fluctuation model (36), where the indole is rotated around the Ca-Cβ and Cβ-Cy bonds by $\phi_1$ and $\phi_2$, respectively. The distribution of $\phi_1$ and $\phi_2$ was a normalized Gaussian function $g_1$ and $g_2$ with a standard deviation of $\sigma_1$ and $\sigma_2$, with a cutoff at $2\sigma_c$. The most probable rotation angle $\phi_{v}$, which corresponds to the center of the Gaussian, was the equilibrium torsion angle $\chi_v$ determined from the $^{13}$C-19F and 19F-19F distance measurements. The two torsional rotations were assumed to be independent, and all covalent bond angles were held fixed.

The Cβ-Cy axis was rotated by a certain angle $\phi_{\lambda,n}$ around the Ca-Cβ bond, then the C-H dipolar tensor in the indole ring was rotated around the Cβ-Cy axis by $\phi_{\lambda,k}$. The motionally averaged dipolar tensor $\overline{D}(\phi_{\lambda,n}, \phi_{\lambda,k})$ was calculated as the weighted average of the individual rotated tensors (3 × 3 matrices) for systematically varied $\phi_{\lambda,n}$ and $\phi_{\lambda,k}$ values, where the weighting function is the product of the normalized Gaussian distributions:

$$\overline{D}(\sigma_1, \sigma_2) = \sum_k \sum_n \overline{D}(\phi_{\lambda,n}, \phi_{\lambda,k}) g_1(\phi_{\lambda,n}, \sigma_1) g_2(\phi_{\lambda,k}, \sigma_2).$$

The principal values $\overline{D}_{\sigma_1}, \overline{D}_{\sigma_2}, \overline{D}_{\sigma_3}$ of the traceless average tensor $\overline{D}$, ordered such that $\overline{D}_{\sigma_1}$ has the largest magnitude and $\overline{D}_{\sigma_3}$ the smallest, give the motionally averaged dipolar coupling constant $\overline{D}$. The asymmetry parameter of the averaged coupling tensor is $\overline{\eta} = (\overline{D}_{\sigma_1} - \overline{D}_{\sigma_3})/\overline{D}_{\sigma_2}$. The calculated $\overline{D}$ was converted to the order parameter as $S_{\text{CH}} = \overline{D}/\overline{D}_{\sigma_2}$, and then compared with the measured $S_{\text{CH}}$ from the DIPSSHIFT experiments. To determine the values of $\sigma_1$ and $\sigma_2$ that are compatible with all measured order parameters, $S_{\text{CH}}$ as a function of $\sigma_1$ and $\sigma_2$ for several indole bonds were superimposed, and the regions consistent with the experimental values were highlighted.

RESULTS AND DISCUSSION

Conformation of Trp$^{41}$

To determine the side-chain conformation of Trp$^{41}$ and its dependence on pH, we measured the Trp$^{41}$ $^{13}$C and $^{15}$N chemical shifts using 2D $^{13}$C-$^{13}$C and $^{15}$N-$^{13}$C correlation experiments. Fig. 2 shows representative 1D and 2D spectra of LW-M2TM. The 1D $^{13}$C spectra of M2TM show similar $^{13}$C intensity patterns at high and low pH. No significant intensity changes were observed between 303 and 243 K, consistent with the previously reported immobilization of M2TM by the cholesterol-rich virus-mimetic membranes (19,20). The Ca and Cβ chemical shifts of L40 and W41 are generally consistent with the α-helical secondary structure, as expected for this four-helix bundle. However, the L40 Ca and Cβ chemical shifts change to more ideal α-helical values at low pH (larger Ca and smaller Cβ chemical shifts), consistent with previous reports that the helix becomes straighter and more tilted at low pH due to charge repulsion at His$^{37}$ (32).

Full assignment of the indole $^{13}$C and $^{15}$N chemical shifts were obtained from 2D $^{13}$C-$^{13}$C DQF correlation spectra and $^{15}$N-$^{13}$C correlation spectra (Fig. 2, c and d). In these spectra, only a single chemical shift was observed for each site, with an average $^{13}$C line-width of 1.8 ppm. This contrasts with the peak doubling reported for the longer M2 construct that contains both the TM and cytoplasmic helices (37,38), for which chemical shift differences as large as 2.7 ppm were detected, which is larger than the resolution limit of our spectra. The chemical shift difference for the peak-doubled longer M2 construct is the largest at backbone Ca (about 2 ppm), which strongly suggests that the cytoplasmic helix affects the backbone conformation of the TM domain (37,38). This effect is consistent with the recent report that the cytoplasmic helix has the ability to cause the formation of high-curvature membrane domains, which shifts the conformational equilibrium of the TM segment to a form that is incompetent for binding the antiviral drug amantadine (19,39). The absence of such peak doubling for the TM construct used here indicates that the TM segment adopts a single conformation in the absence of the cytoplasmic helix. Based on all available experimental evidence so far (6,19,32), this conformation is sensitive to both pH and drug and is thus functionally relevant.

Further information about the local environment of Trp$^{41}$ can be gleaned by comparing the $^{13}$C and $^{15}$N chemical shifts between pH 8.5 and pH 4.5 and between the short and long M2 constructs at high pH (Fig. 2 e). For the TM construct, most side-chain atoms exhibit similar (<0.5 ppm) chemical shifts at high and low pH, with the exception of Cβ1 and Cβ2, which also show nonnegligible chemical shift differences between the short and long M2 constructs (Fig. 2 e). For Ca and Cβ chemical shifts, which are sensitive to the backbone conformation, the long-peptide chemical shifts are closer to the short peptide’s low-pH chemical shifts. This phenomenon can be understood. The presence of the cytoplasmic helix is known to moderately increase proton conduction (40), which implies that the channel may adopt the acid-activated conformation more readily in the presence of the cytoplasmic helix. The M2(22–62) chemical shifts were also measured at slightly lower pH (pH 7.5) than the high-pH sample here, thus a higher percentage of tetramers should exist in the partially charged states for proton conduction. Finally, the longer M2 construct was studied in the more fluid membranes of DOPC/DOPE (dioleoylphosphatidyl-choline/dioleoylphosphatidyl-ethanolamine) in one
case (38) and DPhPC (diphytanoylphosphatidylcholine) membrane in another (37), which should facilitate the conformational motion that is necessary for the TM helix to adopt the low-pH conformation.

His<sup>37</sup>-Trp<sup>41</sup> and Trp<sup>41</sup>-Trp<sup>41</sup> distances

His<sup>37</sup>-Trp<sup>41</sup> aromatic interactions have been implicated in channel gating at low pH based on resonance Raman data (17). Here we directly measured the distances between these two residues using <sup>13</sup>C-labeled His<sup>37</sup> and 5-<sup>19</sup>F-labeled Trp<sup>41</sup> (Fig. 3 <i>a</i>). The latter corresponds to the H<sub>2</sub> position of the indole ring. The high gyromagnetic ratio of the <sup>19</sup>F spin allows <sup>13</sup>C-<sup>19</sup>F distances to be measured up to ~8 Å (~55 Hz) (23,41). We measured the <sup>13</sup>C-<sup>19</sup>F distances using a variant of the REDOR experiment, where a selective <sup>13</sup>C π-pulse was applied in the middle of the REDOR period to suppress one-bond <sup>13</sup>C-<sup>13</sup>C J-coupling and lengthen the effective <sup>13</sup>C T<sub>2</sub> (24,25). Fig. 3, <i>b</i> and <i>c</i>, shows representative REDOR spectra, where the His<sup>37</sup> C<sub>y</sub>, Cε1, and Cδ2 signals, which are sensitive to pH, are assigned according to published 2D spectra (6). Several MAS frequencies were used in these experiments to avoid resonance overlap between the carbonyl sidebands and the aromatic carbons of interest. The largest intensity differences between the control (<i>S</i> <sub>0</sub>) and dephased (<i>S</i>) REDOR spectra were observed for C<sub>g</sub> and Cδ2 of the low-pH sample, which showed <i>S</i>/<i>S</i> <sub>0</sub> values as low as 0.3 by ~8 ms. The high-pH peptide shows comparatively less REDOR dephasing, indicating longer distances between His<sup>37</sup> and Trp<sup>41</sup>. The difference is the largest for C<sub>g</sub> (Fig. 3 <i>d</i>), whose dephasing corresponds to a two-spin C<sub>g</sub>-F distance of 8.4 Å at pH 8.5 but 6.2 Å at pH 4.5. For Cδ2, only low-pH data could be obtained with sufficient sensitivity, and the REDOR dephasing was fast, corresponding to a short two-spin distance of 6.0 Å.

Complementing the His<sup>37</sup>-Trp<sup>41</sup> <sup>13</sup>C-<sup>19</sup>F distances, we also measured the <sup>19</sup>F-<sup>19</sup>F distances using the <sup>19</sup>F CODEX experiment (see Fig. S1 in the Supporting Material). We have previously measured these distances in DMPC (dimyristoylphosphatidylcholine)-bound M2TM, where the peptide was reconstituted into the membrane using a different protocol (10). Under those conditions, the nearest-neighbor distance between the <sup>19</sup>F spins was 11 ± 1 Å at
both high and low pH. For the virus-mimetic membrane samples prepared by detergent dialysis, the CODEX dephasing curves show detectable differences between high and low pH: the best-fit nearest-neighbor distance is 11.3 Å at high pH and 12.4 Å at low pH, indicating that the inter-Trp separation is enlarged at low pH. This is consistent with most available experimental evidence showing that the C-terminal region of the four-helix bundle is expanded at low pH (12).

**pH-dependent Trp\(^{41}\) rotameric structure and His\(^{37}\)-Trp\(^{41}\) contact**

These \(^{13}\)C-\(^{19}\)F and \(^{19}\)F-\(^{19}\)F distances, together with the known backbone structure of M2TM (42) and the known His\(^{37}\) side-chain conformation (6), allowed us to constrain the Trp\(^{41}\) side-chain conformation. Two PDB structures were used for each pH condition to assess the influence of the backbone conformation on the Trp\(^{41}\) rotamer determination. Interresidue distances were extracted as a function of Trp\(^{41}\) \(\chi_1\) and \(\chi_2\) angles to compare with the measured \(^{13}\)C-\(^{19}\)F and \(^{19}\)F-\(^{19}\)F distances. The two high-pH models (PDB:2KQT and PDB:3LBW) yielded a consistent best-fit rotamer of \((-155^\circ, +125^\circ)\) (Fig. 4), indicating that the rotamer is \(-90^\circ\). In contrast, the two low-pH structures did not give a single consensus rotamer. The solid-state NMR backbone structure (PDB:2KAD) gave a best-fit Trp\(^{41}\) rotamer of \((-155^\circ, 135^\circ)\) whereas the low-pH crystal structure (PDB:3C9J) gave two possible Trp\(^{41}\) rotamers: \((-115^\circ, 45^\circ)\) and \((-145^\circ, -125^\circ)\) when solutions with steric clashes were ruled out. Because the PDB:3C9J backbone shows considerable asymmetry, and it is energetically more costly for Trp\(^{41}\) to undergo a large-angle conformational change than a small-angle change between high and low pH, we favor the \((-155^\circ, 135^\circ)\) rotamer as the low-pH Trp\(^{41}\) conformation. The overall similarity of the \(\chi_2\) angle at high and low pH is consistent with the Raman data (17), whereas the trans-\(\chi_1\) angle is consistent with the previously measured His\(^{37}\) Ne2-Trp\(^{41}\) C\(_y\) distance (13). Other rotamer minima seen in Fig. 4 can be ruled out based on steric conflicts. For example, multiple rotamer minima in PDB:3C9J have steric clashes with either the helix backbone or other side chains in the HxxxW segment (see Fig. S2, a and b). The \((-90^\circ, 180^\circ)\) rotamer in PDB:3LBW and PDB:2KAD causes steric clashes among the four Trp side chains (see Fig. S2 c). The \((95^\circ, -125^\circ)\) rotamer in PDB:3LBW and PDB:2KQT is rarely populated in the protein database (43) and is therefore not considered further (see Fig. S2 d).

Fig. 5, a and b, shows model-dependent fits of the \(^{13}\)C-\(^{19}\)F REDOR and \(^{19}\)F-\(^{19}\)F CODEX data at high and low pH for the best-fit Trp\(^{41}\) rotamers. For the \(^{13}\)C-\(^{19}\)F REDOR data, the five-spin and two-spin simulations both show good agreement with the measured data within experimental uncertainty, indicating that the shortest interhelical distance accounts for the majority of the observed REDOR dephasing.

The high- and low-pH His-Trp side chains are compared in Fig. 5, c and d. At both pH conditions, the Trp\(^{15}\) side chain broadly adopts the \(90^\circ\) rotamer, which is consistent with the observed pH sensitivity of the \(\delta\)1C and \(\delta\)2C chemical shifts, because these two carbons of the indole ring face the aqueous pore. In contrast, the \(t\)-105 rotamer would point \(\delta\)1 to the lipid (Fig. 1), inconsistent with the pH sensitivity.
of its chemical shift. Under this broad similarity of the Trp^{41} rotamer, His^{37} and Trp^{41} establish closer contact at low pH, due to a downward movement of the imidazole rings as a result of the larger helix tilt angle. The 15–20° Trp^{41} χ_1 and χ_2 angle differences between the two pH have a more subtle effect on the aromatic packing: the χ_1 change (from −175° at high pH to −155° at low pH) pushes the indole ring toward the center of the channel, while the χ_2 change (from 120° at high pH to 135° at low pH) moves the Cη2/Cξ3 end of the indole toward the C-terminus. These counterdirectional χ_1 and χ_2 changes may serve to avoid steric clashes with the side chains of Leu^{38} and Ile^{42} while still allowing the indole face to be exposed to His^{37}. The closer contact between the imidazole and indole rings of two neighboring helices at low pH creates stronger aromatic interactions, consistent with the Raman spectral changes (17).

**Dynamics of the Trp^{41} side chain**

To investigate whether the Trp^{41} side chain is dynamic to affect proton gating and interaction with His^{37}, we measured the order parameters of several indole 13C-1H and 13C-15N bonds at high temperature. Fig. 6 shows representative 13C-1H and 13C-15N dipolar coupling data obtained at 303 K. The 13C-1H dipolar couplings were measured using the 2D DIPSHIFT experiment whereas the 13C-15N dipolar couplings were measured using REDOR. We found order parameters of 0.7–0.9 for the indole bonds (with uncertainty of 0.05), with the lowest bond order parameters along the C2–C3 and C2–C3 axis. pH change caused the largest difference in the order parameters of the Cα-H and Cζ3-H bonds, with the low-pH sample showing smaller S_{CH} values or larger-amplitude dynamics.

Qualitatively, the relatively large order parameters (>0.7) indicate that the motional amplitude of the indole ring is small at both pH conditions. To determine the motional geometry, we considered simple two-site jump motions around the Cα axis but variable angles from the Cα-Cβ axis depending on the χ_2 angle. The equilibrium values are known from the above His-Trp distance measurements: (−175°, 120°) at high pH and (−155°, 135°) at low pH. Thus, the indole bond orientations relative to the Cα-Cβ and Cβ-Cγ axes are determined by the equilibrium conformation and the jump angle. We calculated the average dipolar couplings as a function of jump angle for the various indole bonds, as shown before for His^{37} (6), and did not find any jump angle that simultaneously satisfied all the measured order parameters. This indicates that the indole ring does not undergo a simple two-site jump motion.

This prompted us to consider a more general motional model, which is Gaussian fluctuation of the indole ring around both the Cα-Cβ and Cβ-Cγ bonds with Gaussian widths σ_1 and σ_2. Successive rotations around the Cα-Cβ and Cβ-Cγ bonds change the orientations of the indole bonds and the associated dipolar couplings. The average dipolar coupling tensors were obtained as a function of the standard deviation, σ_1 and σ_2, of the Gaussian functions. The principal values of the average tensor yielded the order parameter and average asymmetry parameter. Fig. 6, e and f, shows the (σ_1, σ_2) values for the calculated

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**FIGURE 4** Total RMSD between the measured and modeled Ca-F, Cγ-F, Ce1-F, Cε2-F, and 1H-F,19F distances as a function of Trp^{41} (χ_1, χ_2) torsion angles. Different backbone structure models for M2TM at high (a and b) and low (c and d) pH are used for structural modeling. Only the shortest of the four possible distances in the tetramer for each two-spin combination is used to compare with the experimental data. (a) PDB:3LBW backbone structure. (b) PDB:2KQT backbone structure. (c) PDB:3C9J backbone structure. (d) PDB:2KAD backbone structure. Between the two high-pH backbone structures, consensus (χ_1, χ_2) results were found at (−175°, 120°). For the low-pH state, the best-fit Trp^{41} rotamer depends on the backbone structure. For the PDB:2KAD structure (d), the best-fit solution is (χ_1, χ_2) = (−155°, 135°), whereas the PDB:3C9J structure (c) shows two rotamer minima without steric conflict. (Crosses) Rotamers with steric clashes, which are not considered further (see Fig. S2).
Cδ1-Hδ1, Cζ2-Hζ2, Cε3-Hε3, and Cζ3-Hζ3 order parameters that agree with the measured $S_{\text{CH}}$ values within experimental uncertainty. For the low-pH data, the Gaussian widths that agree with all measured $S_{\text{CH}}$ values are approximately $\sigma_1 = 30^\circ$ and $\sigma_2 = 15^\circ$, indicating that the indole undergoes small-angle fluctuations around both the $\chi_1$ and $\chi_2$ bonds. In contrast, at high pH, the larger $S_{\text{CH}}$ for Cδ1-H and Cζ3-H bonds indicate a clear change in the motional geometry, with negligible $\chi_2$ changes. The remaining $\chi_1$ torsional motion has a $\sigma_1$ of 25°. At both pH, the calculations show that the motionally averaged asymmetry parameters do not deviate significantly from uniaxiality (see Fig. S3), thus validating the uniaxial approximation in the order parameter extraction from the experimental data.

Fig. 7 depicts the equilibrium Trp$^{41}$ rotameric conformation at high and low pH as well as their dynamic changes. To indicate the range of excursion of the indole ring, we show two limiting rotamers that are one standard deviation ($\sigma$) away from the equilibrium $\chi_1$ and $\chi_2$ angles. It can be seen that at acidic pH, not only is the average position of the indole ring closer to the imidazolium compared to high pH, but the $\chi_1$ and $\chi_2$ motions further appose the indole to His$^{37}$. As a result, at low pH, the indole alternates between very strong and very weak cation-π interactions with the imidazolium during its reorientational motion (44). To give a quantitative estimate of the His$^{37}$-Trp$^{41}$ proximity, we calculated the distances of His$^{37}$ Nε to the center of the benzene ring and the center of the five-membered pyrrole ring. For the equilibrium low-pH Trp$^{41}$ rotamer of ($-155^\circ$, $135^\circ$), Nε2 is 4.3 Å from the center of the benzene ring and 3.0 Å from the pyrrole ring (Table 2). Due to the torsional fluctuation, the closest approach of indole to the imidazolium has distances of 3.5 Å and 2.5 Å, whereas the remote position of the indole shows a Nε2-benzene distance of 5.3 Å and an Nε2-pyrrole distance of 3.6 Å. For comparison, at high pH, the His-Trp distances for the equilibrium Trp$^{41}$ conformation are longer.
5.4 Å and 3.7 Å, respectively, comparable to the largest separation at low pH, and the Trp41 motional range is too small to make a significant difference (Table 2).

**TABLE 2** Trp41-His37 distances as a function of pH and rotameric dynamics

<table>
<thead>
<tr>
<th>pH</th>
<th>Indole position</th>
<th>Trp41 (χ1, χ2) (A)</th>
<th>Nr2-benzene (A)</th>
<th>Nr2-pyrole (A)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Low</td>
<td>Position 3</td>
<td>(−125°, 150°)</td>
<td>3.5</td>
<td>2.5</td>
</tr>
<tr>
<td></td>
<td>Equilibrium</td>
<td>(−155°, 135°)</td>
<td>4.3</td>
<td>3.0</td>
</tr>
<tr>
<td></td>
<td>Position 1</td>
<td>(175°, 120°)</td>
<td>5.3</td>
<td>3.6</td>
</tr>
<tr>
<td>High</td>
<td>Position 3</td>
<td>(160°, 120°)</td>
<td>5.3</td>
<td>3.6</td>
</tr>
<tr>
<td></td>
<td>Equilibrium</td>
<td>(−175°, 120°)</td>
<td>5.4</td>
<td>3.7</td>
</tr>
<tr>
<td></td>
<td>Position 1</td>
<td>(−150°, 120°)</td>
<td>5.5</td>
<td>3.8</td>
</tr>
</tbody>
</table>

His37 Nr2 distances to the center of the six-membered benzene ring and to the midpoint of the Ne1-Cγ vector of the pyrrole ring are listed. Positions 3 and 1 correspond to those shown in Fig. 7.

**Implication of the Trp41 structure and dynamics to channel gating**

The above results provide direct experimental evidence for the closer contact between His37 and Trp41 at acidic pH and the existence of the cation-π interaction. The closest approach between His37 Ne2 and the indole face is 2.5 Å, which is shorter than the reported distance range of 3.5–4.0 Å between cationic amino-acid side chains such as...
Arg and aromatic rings (44) in proteins. This close proximity is achieved by the combined actions of the increased helix tilt angle and the Trp$^{41}$ side-chain conformational dynamics. At low pH, the equilibrium conformation of Trp$^{41}$ is roughly perpendicular to the His$^{37}$ side chain (Fig. 7), suggesting that the His-Trp interaction has a significant hydrogen-bonding component (44). At high pH, the equilibrium position of the indole ring is similar to the furthest position of the indole from the imidazolium at low pH, and the motional amplitude is insufficient to bring the indole closer to His$^{37}$. Therefore, the aromatic interaction is significantly weaker.

We propose that this close contact between His$^{37}$ and Trp$^{41}$ at low pH is an important reason for the low proton flux of the M2 channel compared to the measured His$^{37}$-water proton-exchange rate. $^{15}$N NMR lineshapes indicated a water-His$^{37}$ proton exchange rate of $4.5 \times 10^6 \text{s}^{-1}$ between pH 5 and 6 (7). This is at least two orders of magnitude faster than the proton flux of $10^{-3} \text{m} \text{s}^{-1}$ per second estimated from whole-cell currents and liposome assays (45–47). Therefore, at most only 1% of His-exchanged protons successfully reach the virion. The dynamically fluctuating His-Trp separation at low pH obtained from our experiments suggest that, when the indole approaches the imidazolium ring, the cation-π interaction should prevent proton dissociation from His$^{37}$ to water, while as the indole moves away from the imidazolium, the protons can be released to water. This concept has been recently proposed based on molecular modeling of the His-Trp dyad (18). Our results provide the first concrete experimental evidence for this dynamic gating model.

In addition to Trp$^{41}$-mediated regulation of the proton flux, other mechanisms can exist to affect the number of protons released to the C-terminus: these include futile exchange between multiple histidines through intervening water molecules (8) and occasional proton release to the N-terminus, although the latter should be much less frequent than proton release to the C-terminus due to the inherent pH gradient in the channel (48).

In a channel with a reverse proton gradient, where the virus interior is acidic but the exterior is neutral, the lack of reverse current has been attributed to the gating function of Trp$^{41}$ (9). Although NMR experiments, similar to most other biophysical techniques, cannot easily mimic the pH gradient used in functional assays, our data provide some insight into this inward rectification. The distance-based structural modeling here indicates that the key first event that reduces the His-Trp separation is the backbone orientational change due to His-His charge repulsion. Thus, as long as the protons from the virus interior cannot pass Trp$^{41}$ to reach His$^{37}$, no imidazole protonation and charge repulsion can occur, and the helix tilt angle will remain relatively small, corresponding to the situation in Fig. 7b.

**CONCLUSION**

The His-Trp distances, Trp$^{41}$ chemical shifts and dynamics measured here provide detailed insights into the aromatic interaction between these two essential functional residues of the M2 proton channel. Our results indicate that the two aromatic side chains are able to approach each other at low pH due to the increased tilt angle of the helix and the microsecond ($\chi_1, \chi_2$) torsional fluctuations of the indole ring at low pH. The low-pH side-chain motion periodically enhances and weakens the His$^{37}$-Trp$^{41}$ cation-π interaction as the indole moves through its trajectory, thus blocking and releasing protons. In contrast, at high pH only $\chi_1$ motion is present and the average position of the indole is further from the imidazole due to small ($\chi_1, \chi_2$) changes. The pH-dependent Trp$^{41}$ side-chain dynamics, with amplitudes of 15–30°, are reminiscent of the small-angle (45°) $\chi_2$ change of His$^{37}$ that shuttles protons (6,7). Nature seems to have designed the compact functional heart of this ion channel by relying on economical small-angle side-chain conformational dynamics, coupled with backbone orientational changes, to regulate proton conduction and channel gating.

**SUPPORTING MATERIAL**

Three figures are available at [http://www.biophysj.org/biophysj/supplemental/S0006-3495(13)00326-3](http://www.biophysj.org/biophysj/supplemental/S0006-3495(13)00326-3).

This work is supported by National Institutes of Health grant No. GM088204.

**REFERENCES**


Supporting Information

pH-Dependent Conformation, Dynamics, and Aromatic Interaction of the Gating Tryptophan Residue of the Influenza M2 Proton Channel from Solid-State NMR

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Figure S1. $^{19}$F CODEX spin diffusion data of 5-$^{19}$F-labeled Trp41 in VM-bound M2TM. (a) Representative CODEX control ($S_0$) and dephased (S) spectra of the pH 8.5 and pH 4.5 samples under 8 kHz MAS. The spectra were measured at 200 K and 235 K without significant intensity differences. The mixing times were (b) Normalized S/$S_0$ intensity as a function of mixing time. The high-pH sample shows faster decay than the low-pH sample, indicating shorter $^{19}$F-$^{19}$F intermolecular distances. Best fits assuming a symmetric tetramer gave a nearest-neighbor distance of 11.3 Å for the high-pH sample and 12.4 Å for the low-pH sample.
Figure S2. Examples of Trp41 ($\chi_1$, $\chi_2$) rotamers that are ruled out because of steric clash or low statistical probability. The $\mathrm{F}^{19}$ position ($\mathrm{H}\zeta_3$) is highlighted in orange. (a) 3C9J backbone with a Trp rotamer of (110°, -45°) has steric clash between the Trp41 $\mathrm{C}\delta_1$ and the helix backbone (red circles). (b) 3C9J backbone with a Trp rotamer of (-60°, 135°) has intrahelical steric clashes between the His37 $\mathrm{Ne}_2$-$\mathrm{H}$, $\mathrm{C}\delta_2$-$\mathrm{H}$ bonds and the Trp 41 $\mathrm{C}\zeta_2$-$\mathrm{H}$, $\mathrm{Ne}_1$-$\mathrm{H}$ bonds. For clarity only two of the four rings are shown for each residue. (c) 3LBW backbone with a Trp41 rotamer of (-90°, 180°) shows Trp-Trp clashes between $\mathrm{Ne}_1$ and $\mathrm{C}\delta_1$, and between $\mathrm{C}\beta$ and $\mathrm{C}\zeta_2$ of adjacent indole rings (red circles). (d) 2KQT backbone with a Trp rotamer (95°,-125°). This is rarely populated in $\alpha$-helices (~2%) (43).
Figure S3. Motionally averaged asymmetry parameters $\overline{\eta}$ as a function of the standard deviations of the Gaussian fluctuations around the Cα-Cβ and Cβ-Cγ axes. Only asymmetry parameters between 0.1 and 0.4 are indicated for simplicity. (a) Low pH data, calculated using $\chi^2=+135^\circ$. (b) High pH data, calculated using $\chi^2=+120^\circ$. The position of the best-fit motional amplitudes at each pH is indicated with a cross. With these motional amplitudes, the averaged asymmetry parameters do not deviate significantly from 1. The Cδ1-H and Cζ3-H couplings have an $\overline{\eta}$ of less than 0.1 at low pH and about 0.2 at high pH. The Cη2/Cε3-H couplings have a modestly larger $\overline{\eta}$ of 0.3. Within experimental uncertainty, these small $\overline{\eta}$ values are not detected in the C-H dipolar dephasing curves, and the uniaxial approximation is reasonable for estimating the order parameters from the experimental data.
The Functional Heart of the M2 Channel

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The equilibration of pH between external environment and viral interior is an essential part of the influenza virus lifecycle. After infection of a eukaryotic cell, the virus needs to transmit protons across its lipidic envelope from the acidic exterior, an endosome, to the interior, eventually triggering unpacking of the viral genome (1). This essential function is fulfilled by the viral protein M2 channel, a tetrameric assembly of four transmembrane helices. The proton pore is formed in the center of the four helices. Two amino-acid residues in the pore are crucial for function, forming a HxxxW motif: Histidine 37 provides for the selectivity and Tryptophan 41 for the pH-dependent gating. Structure, conformation, and mechanism of the M2 channel and in particular of these two residues are subject to intense ongoing research (2).

In this issue of Biophysical Journal, Williams et al. (3) use solid-state NMR spectroscopy to determine the orientation of the Trp41 side chain in the low and high pH states of the M2 channel. So far, 10 different structures have been deposited in the Protein Data Bank database, determined by different techniques: solid-state NMR spectroscopy, solution NMR spectroscopy, and x-ray crystallography (4–13). These 10 structures feature, however, a total of three different rotamers for the Trp41 chain. Williams et al. (3) now employ solid-state NMR distance measurements between selected nuclei in specifically isotope-labeled samples to distinguish between the rotameric Trp41 orientations. Importantly, the channel is reconstituted in native-like lipid bilayers for the solid-state magic-angle spinning NMR experiments.

The proton conductance mechanism of the M2 channel had previously been determined. In the high pH closed state, the four His37 side chains block the channel by forming a π-stacked structure. At low pH, however, the His37 cluster is conductive, due to fast protonation and deprotonation as part of a proton wire (14,15). Together with these previous experiments, the orientation of the Trp side chain can now further explain the conducting and gating mechanism of the M2. Trp41 is found to adopt a 90° rotamer in both pH states, with 20° difference in χ1 and χ2 between the high and low pH states. Additional measurements determine the amplitudes of the Trp side-chain dynamics, showing increased fluctuations at low pH. Importantly, His37 and Trp41 approach each other closely at low pH by a cation-π interaction. This close interaction provides the structural basis for the gating by Trp, as it reduces the proton dissociation rate from His37 to water from 4.5 × 105 s⁻¹ to <1% thereof. Furthermore, the Trp gate, which is located toward the virus interior, blocks the accessibility of protons to the His gate and thus results in the overall unidirectional proton conductance. Overall, the visualization of the measured amplitudes of the two side chains pumping protons through the M2 channel by their reorientation evokes the visual analogy of a stroking heart that pumps blood through its valves (Fig. 7 of Williams et al. (3)).

Determination of amino-acid side-chain rotamers is a fundamental biophysical challenge, often necessary to understand protein function at atomic detail. Specific solid-state NMR experiments, such as the REDOR (16) and CODEX (17) experiments used here, in combination with suitable isotope labeling schemes are central techniques in the ongoing work of Luo and Hong (18), Luo et al. (19), Hong et al. (20), and Wang et al. (21). This article demonstrates the potential of solid-state NMR to determine conformation and dynamics of amino-acid side chains at atomic resolution in the functional center of a membrane protein, outlining a procedure to be carried out also on larger and more complex membrane protein systems.

References


http://dx.doi.org/10.1016/j.bpj.2013.03.020


